

#### Max. Marks: 60

# JB 2 MR BATCH PHYSICS : DCT Topic: NLM + Work Energy Power

- 1. A body is at rest under the action of three forces, two of which are  $\vec{F_1} = 4\hat{i}, \vec{F_2} = 6\hat{j}$ . The third force is:
  - (a)  $4\hat{i} + 6\hat{j}$  (b)  $4\hat{i} 6\hat{j}$  (c)  $-4\hat{i} + 6\hat{j}$  (d)  $-4\hat{i} 6\hat{j}$
- 2. The time taken by a block of wood, initially at rest to slide down a smooth inclined plane 9.8 m long (angle of inclination =  $30^\circ$ ) is:



3. A wooden block is placed on an inclined plane. The block just begins to slide down when the angle of the inclination is increased to 45°. What is the coefficient of friction?

- (a) 0.25 (b) 0.75 (c) 1 (d) 0.5
- 4. A block pressed against a vertical wall is in equilibrium. The minimum coefficient of friction is:



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- 5. A man wants to remain in equilibrium by pushing his hands and feet against two vertical parallel walls as shown in the figure.
  - A. He must exert equal forces on both walls
  - B. The forces of friction at both walls must be equal
  - C. The coefficients of friction between man and wall must be the same at both ends
  - D. Friction must be present on both walls



- (a) A and B are correct
  (b) A and C are correct
  (c) A and D are correct
  (d) All correct
- 6. Three blocks of equal masses (each 3 kg) are suspended by weightless strings as shown. If the applied force is 100 N, then is equal to:  $(g = 10 \text{ m/s}^2)$





7. A horizontal force of 10 N is necessary to just hold a block stationary against a wall. The coefficient of friction between the block and the wall is 0.2 The weight of the block is



8. Three blocks of masses 1 kg, 4 kg and 2 kg are placed on a smooth horizontal surface. If shown in the figure. Two horizontal forces 120 N and 50 N are applies on the system the acceleration of the system is



9. The rear side of a truck is open and a box of 40 kg mass is placed 5 m away from the open end as shown in the figure. The coefficient of friction between the box and the surface below itis 0.15. On a straight road the truck starts from rest and accelerates with 2 ms<sup>-2</sup>. The distance travelled by the truck atthe time the box fall from it is (Ignore the size of the box)





(a)

2.0 kg

(b)

4.0 kg

10. The blocks A and B are arranged as shown in the figure. The pulley is frictionless. The mass of A is 10 kg. The coefficient of friction of A with the horizontal surface is 0.20. The minimum mass of B to start the motion will be



11. The coefficient of static friction, between block A of mass 2kg and the table as shown in the figure is 0.2. What would be the maximum mass value of block B so that the two blocks do not move? The string and the pulley are assumed to be smooth and massless.  $(g = 10 \text{ m/s}^2)$ 



(d) 0.4 kg

12. A 50 kg person stands on a 25 kg platform. He pulls on the rope which is attached to the platform via the frictionless pulleys as shown in the figure. The platform moves upward at a steady rate if the force with which the person pulls the rope is:





13. Mass m rests on a horizontal surface. The coefficient of friction between the mass and the surface is  $\mu$ . If the mass is pulled by a force F as shown in the figure, the limiting friction between the mass and the surface will be:



14. Two block each of mass M are resting on a frictionless inclined plane as shown in the figure. Then:



- (a) The block A moves down the plane (b) The block B moves down the plane
- (c) Both blocks remain at rest
- 15. An insect crawls up a hemispherical surface very slowly (see the figure). The coefficient of friction between the insect and the surface is 1/3. If the line joining the centre of the hemispherical surface to the insect makes an angle with the vertical, the maximum possible value of  $\alpha$  is given by

(d)

Both the blocks move down the plane





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# JB 2 MR BATCH

# **CHEMISTRY: DCT**

# **Topic: Structure of Atoms + Periodic + Mole Concept**

- 16. The number of nucleons in chlorine-37 is
  - (a) 17 (b) 20 (c) 54 (d) 37
- 17. What is the work function of the metal if the light of wavelength 4000 A generates photoelectrons of velocity  $6 \times 10^5 \text{ms}^{-1}$  form it? (Mass of electron =  $9 \times 10^{-31}$ kg, Velocity of light =  $3 \times 10^8 \text{ms}^{-1}$ , Planck's constant =  $6.626 \times 10^{-34}$ Js, Charge of

(Mass of electron =  $9 \times 10^{-19}$  kg, velocity of light =  $3 \times 10^{-19}$  ms , Planck's constant =  $6.626 \times 10^{-19}$  Js, Charge of electron =  $1.6 \times 10^{-19}$  JeV<sup>-1</sup>)

- (a) 0.9 eV (b) 4.0 eV (c) 2.1 eV (d) 3.1 eV
- 18. The size of nucleus is of the order of
  - (a)  $10^{-12}$  m (b)  $10^{-8}$  m (c)  $10^{-15}$  m (d)  $10^{-10}$  m
- 19. The number of neutrons in the dipositive zinc ion (Mass no. of Zn=65)
  - (a) 35 (b) 33 (c) 65 (d) 67

### 20. A metal surface is exposed to solar radiations

- (a) The emitted electrons have energy less than a maximum value of energy depending upon frequency of incident radiations
- (b) The emitted electrons have energy less than maximum value of energy depending upon intensity of incident radiation
- (c) The emitted electrons have zero energy
- (d) The emitted electrons have energy equal to energy of photos of incident light
- 21. The number of unpaired electrons in the  $Fe^{3+}$  ion (atomic no. = 26) is
  - (a) 5 (b) 6 (c) 2 (d) 4
- 22. Uncertainty in the position of an electron (mass=  $9.1 \times 10^{-31}$ kg) moving with a velocity 300 m/s, accurate up to 0.001% will be: (h =  $6.63 \times 10^{-34}$  Js)

(a)  $19.2 \times 10^{-2}$  m (b)  $5.76 \times 10^{-2}$  m (c)  $1.92 \times 10^{-2}$  m (d)  $3.84 \times 10^{-2}$  m



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| 23.  | To whi  | ch of the following   | is Bohr'   | s theory applicable  |  |   |                   |                       |  |  |
|--|---|---|--|--|--|---|-------------------|-----------------------|--|--|
|  | (a)   | He <sup>+</sup>   | (b)  | Li <sup>+2</sup>   | (c)                                      | Tritium   | (d)               | Be <sup>+2</sup>      |  |  |
|  | (a)   | III, IV   | (b)  | I,II,III,IV  | (c)                                      | I,II  | (d)               | I,II,III              |  |  |
| 24.  | Find th   | e value of oxidation  | state of   | $[Co in Ag(Co(CO)_4]]$   | :  |   |                   |                       |  |  |
|  | (a)   | 1   | (b)  | -1   | (c)                                      | Zero  | (d)               | None of these         |  |  |
| 25.  | Which   | of the following rea  | ctions in  | nvolve oxidation and   | reductio                                 | n?  |                   |                       |  |  |
|  | (a)   | $NaBr + HCl \rightarrow NaCl + HBr$   |  |  |  | $HBr + AgNO_3 \rightarrow AgBr + HNO_3$           |                   |                       |  |  |
|  | (c)   | $\mathrm{H}_2 + \mathrm{Br}_2 \rightarrow 2 \ \mathrm{HBr}$   |  |  | (d)                                      | $Na_2O + H_2SO_4 \rightarrow Na_2SO_4 + H_2O$     |                   |                       |  |  |
| 26.  | Manganese achieves its maximum oxidation state in its compound:   |   |  |  |  |   |                   |                       |  |  |
|  | (a)   | $MnO_2$   | (b)  | $Mn_3O_4$  | (c)                                      | KMnO <sub>4</sub>                                 | (d)               | $K_2MnO_4$            |  |  |
| 27.<br>28.   | Which<br>(a)<br>(b)<br>(c)<br>(d)<br>The inc<br>(a)<br>(b)<br>(c)<br>(d)                                      | statement is wrong<br>Oxidation number<br>Oxidation number<br>Oxidation number<br>Oxidation number<br>Oxidation number<br>correct order of decr<br>$H_2S_2O_7 > Na_2S_4O_6$<br>$H_2SO_5 > H_2SO_3 >$<br>$SO_3 > SO_2 > S_8 > 1$<br>$H_2SO_4 > SO_2 > H_2$ | of oxyg<br>of oxyg<br>of oxyg<br>of oxyg<br>easing c<br>$> Na_2S$<br>$SCl_2 > H$<br>$H_2S$<br>$S > H_2S$ | en is +1 in peroxides<br>en is +2 in oxygen di<br>en is $-\frac{1}{2}$ in superoxid<br>en is (-2) in most of i<br>pxidation number of S<br>${}_{2}O_{3} > S_{8}$<br>${}_{2}S$<br>${}_{2}O_{8}$ | fluoride<br>les<br>ts compo<br>5 in comj | ounds<br>pound is :                               |                   |                       |  |  |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | The reaction $3\text{CIO}^{-}(aq) \rightarrow \text{CIO}_{3}^{-}(aq) + 2\text{CI}^{-}(aq)$ is an example of : |   |  |  |  |   |                   |                       |  |  |
|  | (a)   | oxidation   |  |  | (b)                                      | reduction   | _                 |                       |  |  |
|  | (c)   | disproportionation  |  |  | (d)                                      | decomposition react                               | tion              |                       |  |  |
| 30.  | The ox<br>(a)<br>(b)<br>(c)<br>(d)  | idation number of a<br>actual charge of th<br>valency of the ator<br>formal charge of th<br>actual charge of th   | n atom<br>e atom<br>n<br>ne atom<br>ne atom  | in a given species (m<br>if the atom exists as   | olecule,<br>a monoa                      | ion or free atom) is th<br>atomic ion, or the hyp | ne :<br>pothetica | al charge assigned to |  |  |
|  |   | the atom in the spe   | cies by  | simple rules.  |  |   |                   |                       |  |  |



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| 1.  | (d) | 2.  | (b) | 3.  | (c) | 4.  | (c) | 5.  | (c) |
|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| 6.  | (a) | 7.  | (d) | 8.  | (b) | 9.  | (a) | 10. | (a) |
| 11. | (c) | 12. | (b) | 13. | (c) | 14. | (a) | 15. | (a) |

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# JB 2 MR BATCH CHEMISTRY: DCT ANSWER KEY Topic: Structure of Atoms + Periodic + Mole Concept

| 16. | (d) | 17. | (c) | 18. | (c) | 19. | (a) | 20. | (a) |
|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| 21. | (a) | 22. | (c) | 23. | (d) | 24. | (b) | 25. | (c) |
| 26. | (c) | 27. | (a) | 28. | (d) | 29. | (c) | 30. | (d) |